NOTICE CONCERNING COPYRIGHT RESTRICTIONS

This document may contain copyrighted materials. These materials have been made available for use in research, teaching, and private study, but may not be used for any commercial purpose. Users may not otherwise copy, reproduce, retransmit, distribute, publish, commercially exploit or otherwise transfer any material.

The copyright law of the United States (Title 17, United States Code) governs the making of photocopies or other reproductions of copyrighted material.

Under certain conditions specified in the law, libraries and archives are authorized to furnish a photocopy or other reproduction. One of these specific conditions is that the photocopy or reproduction is not to be "used for any purpose other than private study, scholarship, or research." If a user makes a request for, or later uses, a photocopy or reproduction for purposes in excess of "fair use," that user may be liable for copyright infringement.

This institution reserves the right to refuse to accept a copying order if, in its judgment, fulfillment of the order would involve violation of copyright law.

OPTIMIZED H2S TREATMENT AT PACIFIC GAS AND ELECTRIC COMPANY'S GEVSERS GEOTHERMAL PROJECT UNIT 11

Gary P. Dorighi William A. Conner Larry H. Kirby Pacific Gas & Electric Pacific Gas & Electric Dow Chemical U.S.A.

ABSTRACT

The geothermal steam at Pacific Gas and Electric Company's Geysers Geothermal Project contains hazardous constituents of which hydrogen sulfide (H₂S) presents the greatest problem. Hydrogen sulfide removal processes were developed in the mid-1970's to comply with local air pollution control regulations. These chemical processes consumed a large volume of expensive chemicals and generated the major portion of the total waste for the geothermal plant.

This report describes the operation of a new process that was applied at Unit 11. This integrated system uses ferric HEDTA chelate to produce zerovalent sulfur in the condensate and the reduced iron chelate is reoxidized in the cooling tower. Thermal incineration and scrubbing of the non-condensible gas produces sodium bisulfite which is contacted with the zerovalent sulfur to produce soluble thiosulfate which remains in the geothermal condensate.

This process has greatly reduced the consumption of expensive chemicals and virtually eliminated the generation of hazardous wastes at Unit 11. The chemistry and the process results will be discussed.